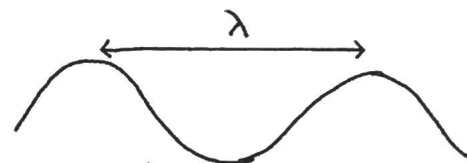


LIGHT

- Light is an electromagnetic wave.
- Two types of waves
 - a) transverse waves - vibrations perpendicular to the direction in which the wave is travelling.
 - b) longitudinal waves - vibrations parallel to the direction in which the wave is travelling.

- Light propagates as transverse waves.
Sound propagates as longitudinal waves.



- A wave is characterized by its wavelength λ (lambda) and frequency f
wavelength \times frequency = velocity

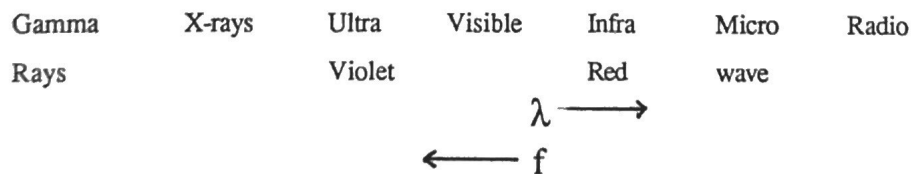
$$\lambda \times f = c$$

- For visible light λ is measured in Angstroms (\AA) $1 \text{ \AA} = 10^{-8} \text{ cm}$
Frequency is measured in Hertz (Hz) : $1 \text{ Hz} = 1 \text{ cycle / sec}$
 $c = \text{speed of light} = 3 \times 10^{10} \text{ cm/sec} = 3 \times 10^8 \text{ m/sec} = 3 \times 10^5 \text{ km/s}$

- Visible spectrum : 4000 \AA to 7000 \AA

4000 \AA	Violet	5500 \AA	Yellow
4500 \AA	Blue	6000 \AA	Orange
5000 \AA	Green	6500 \AA	Red

- Visible light part of the electromagnetic spectrum



Properties of Light

i) Reflection :

angle of incidence = angle of reflection [$i = r$]

ii) Refraction

$\sin i / \sin r = n$ [refractive index]

iii) Diffraction : The ability of light to bend around corners

- Dual nature of light

- wave
- particle

- Photon : A packet of light energy



- Energy of photon : $E \propto f$

$E = h \times f$ $h = \text{Planck's constant}$

- Types of Spectrum [Kirchhoff's Laws]

I. A hot dense gas or solid produces a continuous spectrum.

II. A hot rarefied gas produces emission spectrum.

III. A relatively cool gas in front of a hot, continuous source produces an absorption spectrum. [Fraunhofer Lines]

